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LASER-BASED IGNITION OF H₂/O₂ AND D₂/O₂
PREMIXED GASES NEAR 243 NM:
THE FIRST REPORT OF A DEUTERIUM ISOTOPE
WAVELENGTH EFFECT IN LASER IGNITION

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DECEMBER 1991



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1. INTRODUCTION

The use of lasers to initiate combustion events in reactive gaseous mixtures has been the subject of recent investigations in many laboratories (Forch and Miziolek 1986, 1987; Syage et al. 1988; Lavid and Stevens 1985; Raffel, Warntaz, and Wolfrum 1985). Laser ignition has generally been achieved through photochemical initiation (combustion initiation through chainbranching chemical reactions), thermal heating (laser heating of gases), nonresonant spark formation (which results from gas breakdown by intense laser radiation), and through the resonant multiphoton photochemical formation of microplasmas. Laser-induced gas breakdown, which is defined as the point in which single ionization of the gas occurs, results from the absorption of many photons which leads to ionization, collision-induced cascade ionization, and spark formation. It has been shown, however, that the laser energy at the threshold of nonresonant gas breakdown is typically far in excess of the requisite minimum ignition energy such that the formation of a blast wave may lead to a detonation (Weinberg and Wilson 1971). Here, nonresonant gas breakdown refers to spark formation in a gas which is initially transparent (early on in the laser pulse) at a particular laser wavelength. At sufficiently high field strength and as the laser pulse increases in time, electrons can be generated from multiphoton ionization of the parent fuel and/or oxidizer molecule. Electrons can then absorb radiation from the intense laser field, ionize other gas molecules, and eventually lead to cascade ionization and gas breakdown. It naturally follows that the laser energy at the sharp gas breakdown threshold may indeed exceed the minimum energy required to ignite the gas mixture and thus the extraction of a minimum ignition energy from this experiment may be difficult. However, if the laser energy, threshold for the photoproduction of free electrons can be reduced (and thus electrons are produced early on in the laser pulse) then laser spark ignition may be a more controllable process. The efficient formation and characterization of laser produced sparks (microplasmas) as an ignition source is the subject of this paper.

The resonant multiphoton photochemical formation of microplasmas, which is an ignition means that we have been investigating at the Ballistic Research Laboratory, appears to be a more controllable means to generate laser-produced sparks than gas breakdown. Recent investigations (Forch and Miziolek 1986, 1987) have revealed the first example of a strong wavelength dependence in the amount of incident laser energy which was required to ignite a

premixed flow of H₂/O₂ at atmospheric pressure. A tunable ultraviolet (UV) laser system which operates near 225.6 nm was found to induce photodissociation of the oxidizer component, O₂ or N₂O, to produce oxygen (O) atoms in three ground-electronic spin-orbit split states 2p⁴ ³P_{2,1,0}. It was found that the minima in a piot of incident laser energy (ILE) required for ignition vs. wavelength were located exactly at the same spectral positions as the oxygen atom two-photon allowed absorption transitions from the 2p⁴ ³P_{2,1,0} states to the lowest excited state of the same symmetry (Alden et al. 1982; Miziolek and DeWilde 1984; Meier, Kohse-Hoinghaus, and Just 1986; Goldsmith 1983; Dagdigian, Forch, and Miziolek 1988). Subsequent detailed experimental investigations (Forch and Miziolek 1986, 1987) resulted in the formulation of a mechanism for this process which consists of three components: (1) the multiphoton photochemical formation of oxygen atoms; (2) resonant multiphoton ionization of these atoms to efficiently form free electrons in the laser focal volume early in the laser pulse; (3) and the controlled, resonance-enhanced formation of a microplasma using seed electrons which were generated in the previous process.

This ignition method appears to alleviate the problems associated with the sharp and uncontrolled ignition thresholds which are encountered in the extraction of minimum ignition energy measurements using desirable short-pulse lasers (10° sec) (where energy release occurs within a very short time, over a very small volume in free space and is not associated with catalytic and intrusive effects of electrode surfaces). Therefore, we began a detailed experimental investigation of the potential photochemical interaction of the UV laser and the fuel components of these premixed flows. In this paper, we describe a sensitive wavelength dependence on the laser energy required to ignite a premixed gaseous flow of H₂/O₂ and D₂O₂ through resonant multiphoton excitation of hydrogen (H) and deuterium (D) atoms near 243 nm. We show that there is a definitive excitation wavelength shift near 243 nm (11 cm⁻¹ at the single photon wavelength, 243 nm) for the resonant formation of microplasmas that corresponds exactly to H-D deuterium isotope shift of 22 cm⁻¹ at the two-photon excitation energy (two photons at 143 nm). (D is excited at a wavelength 11 cm⁻¹ to the blue of H and the 2S level of D is 22 cm⁻¹ higher in energy than the 2S level of H). Plots of ILE vs. equivalence ratio, Φ , with the excitation wavelength of the laser tuned to either the H or Datom two-photon transition, shows a minimum at $\Phi = 0.7$ in the fuel lean region at an ILE of ~0.55 mJ. Characterization of the photochemistry involved in microplasma formation was

made. A pressure threshold for microplasma formation was determined, and an estimate of the laser power dependence for the ionization process was made.

2. EXPERIMENTAL

A schematic of the experimental apparatus that was used in this investigation is given in Figure 1. Tunable laser radiation near 243 nm was generated by using the second harmonic (532 nm) of a Q-switched Quanta-Ray Nd:YAG laser (DCR-2A,) to pump a Quanta-Ray Dye laser (PDL-1) which was operated at ~580 nm with a DCM dye. The dye laser beam was then frequency doubled using an angle-cut KDP crystal in the first stage of a servo-motor based tracking system, Quanta-Ray WEX-1; then the doubled radiation was frequency mixed with the 1.06l fundamental from the Nd:YAG in an angle-cut KDP crystal which was contained in the second-stage, WEX-1 module. The 243 nm laser radiation was separated from unwanted beams with two Pellin-Broca prisms (which were positioned to avoid net beam-steering), a broad-band pass filter and aperture. This configuration yielded 7-nsec (10 Hz), ~2.0 cm⁻¹ band width FWHM, 1.5-mJ tunable laser pulses.

The atmospheric pressure burner system consists of a water-cooled and argon-shrouded nozzie with a pinhole aperture (0.6 mm) that was mounted onto an x-y-z translation stage. Gas flows were controlled with Matheson (Model 629) flow meters or Tylan mass flow controllers (Model FC-280-V), which were calibrated by a GCA Precision Scientific wet test meter for H_2 , D_2 and O_2 flows up to 2 L/min. Linear flow velocities were in the 10^3 cm/sec range. The incident laser energy was measured just before a 50-mm focal length lens (the beam waist at the focus was about 50 μ) with a Scientech (Model 38-0103) disc calorimeter-power-energy meter. The output laser energy could be varied by insertion of one (or more) dielectrically-coated partially transmitting filters in the beam and/or attenuation of the Nd:YAG amplifier stage gain. It was found that the beam characteristics (such as spatial and temporal profiles) were unaltered using this method.

Laser ignition measurements were made by flowing a carefully metered, homogeneous mixture of fuel and oxidizer through the burner. The beam was precisely focused ~0.5 mm above the pinhole orifice, then the laser energy was adjusted until a single pulse ignited the gases and a flame stabilized on the burner surface. The flame was immediately extinguished

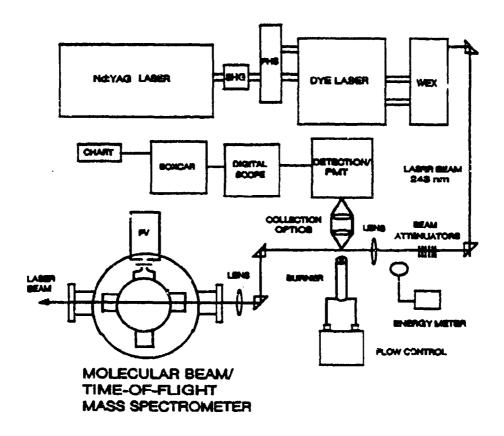


Figure 1. Schematic of Experimental Apparatus.

(fuel and oxidizer valves were closed) in order to maintain a constant burner temperature (as not to preheat the gases in succeeding measurements and to conserve the gases, especially deuterium) then the incident laser energy was measured. It was found that the measured values of ILE for ignition were independent of lateral positioning across the burner aperture (entrainment of room air in the Ai-shrouded, premixed gas jet was not occurring). Wavelength dependent ignition plots near 243 nm were generated by holding U constant and measuring ignition ILE as a function of excitation wavelength.

Excitation scans of the laser produced microplasma emission (from H or D atoms) (in cold gaseous flows or a variable pressure cell) were recorded by collecting emission (with a pair of lenses that were matched to the f/number of a 0.22-m McPherson model 180 monochromator-Hamamatsu R928 photomultiplier system). Emission signals were captured on a 500-MHz Hewlett-Packard model 54111D digital oscilloscope and/or Stanford Research Systems boxcar integrator and strip chart recorder.

Ion signals were detected in a R.M. Jordan time-of-flight mass spectrometer or using a platinum-tipped optogalvanic probe which was biased to detected free electrons liberated in the ionization process. The TOF system consists of a pulsed-molecular Learn valve, a skimmed differentially pumped laser ionization region, a 1.3-m drift tube and microchannel plate detector. Operation pressure is ~10⁻⁷ Torr. The optogalvanic probe was used in atmospheric pressure gas samples. All gases were Matheson UHP and were used "as is" without further purification.

3. RESULTS AND DISCUSSION

3.1 Microplasma Formation. Figure 2 illustrates the single laser spectroscopic schemes which have been utilized for the detection of H atoms in flames, molecular beams, discharges, and other environments (Lucht et al. 1983; Alden et al. 1984; Goldsmith 1982; Goldsmith 1984; Tjossem and Cool 1983; Forch, Morris, and Miziolek 1990). The energy gap between the 1S ground electronic state and the 2S lowest excited electronic state (near-degenerate 2S₁₂₁ 2P₁₂₁ 2P₃₂ levels) is ~10 eV (Figure 2, scheme 1). Since tunable laser sources which operate in this regime are not readily available, two-photon absorption of laser radiation at 243.07 is required to excite the 1S-2S transition. Absorption of an additional third photon from the resonantly excited 2S state is sufficient to ionize an H atom since the three-photon excitation energy (3 x 5.1 eV) exceeds the 13.6 eV H atom ionization potential. (Note that absorption of an additional photon by an H atom in any one of the excited state electronic manifolds given in Figure 2, schemes 1-5, are sufficient to achieve ionization.) Detection of laser-induced fluorescence from resonance excitation of the n=2 level in flames and other environments is difficult since the emission at 122 nm is absorbed by flame and room gases. Therefore, we investigated the possibility of inducing multiphoton photolysis of H₂ in a cold atmospheric flow, followed by resonance enhanced (2 + 1) multiphoton ionization H atoms to form a laser-produced microplasma. Within the high temperature plasma environment, there exist collisional processes which may populate (for example) the n=3 or 4 levels where Balmer- β (486.1 nm) or Balmer- α (656.3 nm) emissions could be detected (monitored by setting the monochromator at the wavelength of interest). Figure 3a gives an excitation wavelength scan where Balmer-β emission was monitored. The signal maximum is found precisely at the known two-photon resonance transition of atomic hydrogen, although the spectral width is much larger than that of a purely atomic transition. This observation is

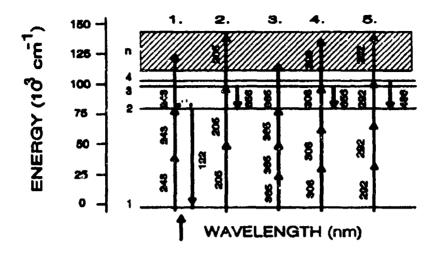


Figure 2. Simple Energy Level Diagram for Atomic Hydrogen. Schemes 1–5 Depict Various Laser-Based Multiphoton Processes for Excitation and Ionization of the H Atom.

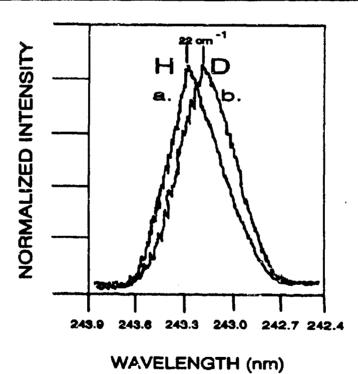


Figure 3. Excitation Wavelength Scans for Microplasma Formation Near 243 nm With the Laser Beam Focused (ILE~1.2 mJ) Into Cold Flows of: (a) H₂, λ obs. = 656.3 nm and (b) D₂, λ obs. = 656.1 nm. The Wavelength Separation Between Peaks Is 11 cm⁻¹ Which Corresponds to 22 cm⁻¹ at the Two-photon Excitation Energy Which Is Identical With the H-D Deuterium Isotope Shift.

clearly indicative of a much more complex, highly nonlinear phenomenon, i.e., the formation of a microplasma. Multiphoton absorption and ionization generates free electrons (priming or seed electrons) early in the laser pulse which initiate cascade ionization and the formation of a plasma which is heated up to a very high temperature by the inverse brehmsstrahlung effect.

The temporal profile of the emission, Figure 4b, is also indicative of microplasma formation. The lifetime of the emission (~40 ns) is much greater than that of the time-resolved scattered laser radiation (~7 nsec) (Figure 4a) or, for example, from resonance multiphoton laser induced fluorescence from H-atoms that we have detected using scheme 5, where the emission signal follows the temporal profile of the exciting laser pulse (Figure 4a). Although microplasma formation is initiated by atomic ionization, the lifetime of the emission is purely a property of the microplasma.

Figure 5a–d, gives the emission profiles as a function of four different laser pulse energies (0.15–1.5 mJ). The widths of the spectral profiles both increases and shows a linear dependence on laser pulse energy. Extrapolation of the measured linewidths to zero laser energy give a FWHM of ~6 cm⁻¹, which is a factor of about six times greater than the measured spectral bandwidth of the laser at this wavelength. It is known, for example, that at pressures near atmospheric the radiation intensities required to cause breakdown by the cascade processes are sufficiently strong that optically induced Stark shifts (which is a linear function of laser energy) and broadening of the electronic levels of an atom occur and have a marked influence on transition probabilities (Tjossem and Cool 1983; Lambropoulos 1974). The signals certainly are severely AC Stark broadened (9 cm⁻¹/GW cm²), however the widths are much greater than the expected natural linewidth of the atomic transition (<<1cni⁻¹). This evidence further indicates that emission we observe is not simply an atomic property, but that of a much more complex event.

To further verify the importance of resonance enhancement in multiphoton photochemical formation of a microplasma, these experiments were repeated under identical conditions except that H_2 was replaced with D_2 . Figure 3b gives the normalized spectral profile of emission signal which was observed. Note that spectra (a) and (b) are essentially identical except that the emission signal in (b) reaches a maximum at a wavelength position which is ~ 11 cm⁻¹ to the blue of (a) (the spacing at the two-photon energy is ~ 22 cm⁻¹). The

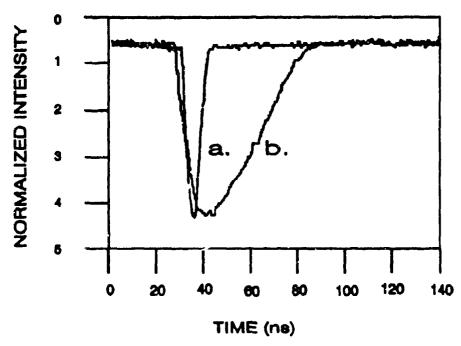


Figure 4. Time Resolved Emission From: (a) Scattered Laser Radiation and (b) H-Atom Emission at 656.3 nm From Microplasmas Formed in H₂ Flows, Using Scheme 1. Figure 2, ILE = 1.2mJ.

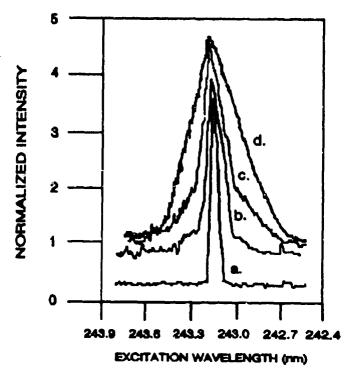


Figure 5. Excitation Wavelength Scans for Microplasma Formation Near 243 nm Plotted as a Function of ILE: (a) 0.15 mJ, (b) 0.43 mJ, (c) 0.73 mJ, and (d) 1.4 mJ in at 300 K in an Atmospheric Pressure Flow.

observation wavelength-bandpass of the monochromator was broadened and centered to equally catch both emissions at 656.3 nm and 656.1 nm for H and D, respectively. A simple calculation and check with a known value 19 shows that this separation is nearly identical with the H-D isotope shift (the D 2S level is 22.4 cm⁻¹ higher in energy than the H 2S level). This isotopic shift is indeed identical to that which has been observed in spectroscopic probes of H and D atoms in flames and other environments. These results strongly support our contention that microplasma formation is the result of initial photolysis of H₂ or D₂ to produce ground electronic state H and D atoms and subsequent multiphoton excitation and ionization are important in the efficient wavelength dependent formation of these microplasmas, just as had been observed previously for oxidizers, O₂ and N₂O (Forch and Miziolek 1986, 1987). The photoproduction of free electrons early on in the temporal evolution of the laser pulse from resonant multiphoton ionization of H or D atoms is responsible for the efficient microplasma formation. When the laser is tuned off the resonance absorption wavelengths of either H or D atoms, then no microplasma formation is seen under these experimental conditions.

3.2 Laser Power Dependencies. In order to gain information on the laser power dependence for the photolysis and mulitphoton ionization processes, we performed a series of experiments in a well-controlled low-pressure (10⁻⁷ Torr) environment using a molecular-beam time-of-flight mass spectrometer (Figure 1). Curiously, we were unable to detect any H* or D* from laser irradiation of molecular beams of H₂ or D₂ when the laser beam was focused into the ionization region of the spectrometer (with the tightest focusing (200-mm f.l. lens) that neometrical constraints would allow). We did, however, find a pressule threshold of ~70 Torr, ILE=1.2 mJ, for the onset of microplasma formation in a variable pressure flow cell. Comparison of these results suggests that a collisionally induced photodissociation process may be responsible for the production of ground state H and D atoms and subsequent microplasma formation, and, is currently under investigation. We were able to generate intense pulsed and effusive (ground state) H and D atom beams with a hot wire tungsten filament inserted before the skin med differentially pumped ionization region in the TOF-M 20. No ions were produced from the hot wire filament itself. Ions were detected only when the laser was tuned on resonance with H and D atoms. Excitation spectra (Figure 6) were recorded by mass gating either the H* or D* signals with a boxcar integrator, then scanning the laser through the two-photon resonance, three photon ionization transitions (2+1) (note

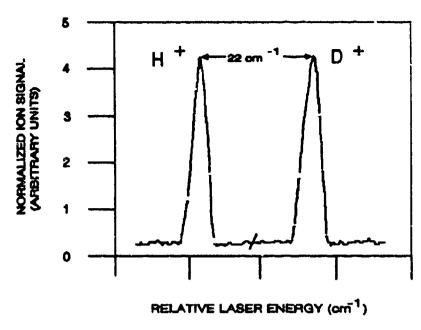


Figure 6. Excitation Spectra of H⁺ and D⁺ Taken in a Time-of-Flight Mass Spectrometer Using Resonance Two-Photon Excitation and Three-Photon Ionization (2+1) of H and D Atoms Near 243 nm. The Wavelength Separation Between Peaks Is 11 cm⁻¹ Which Corresponds to 22 cm⁻¹ at the Two-Photon Excitation Energy Which Is Identical With the H-D Deuterium Isotope Shift.

that there was virtually no signal off resonance of either excitation wavelength). Again, a wavelength spacing of 11 cm⁻¹ is evident, which corresponds to a deuterium isotope shift of 22 cm⁻¹ at the two-photon energy). Laser power dependencies for ion signal production were made assuming a signal dependence on laser intensity of Iⁿ, where n equals the number of photons absorbed. Values of n=2.75+/-0.24 and 2.63+/-0.36 were obtained for H⁺ and D⁺ production, respectively, which corresponds to two-photon resonance excitation and one photon ionization at low iaser energies (<0.5 mJ). These results (*e.s.* expected) indicate that the overall excitation and ionization was a three-photon process. The noninteger values are indicative of partial saturation of the initial two-photon absorption and ionization step. We could easily fully saturate the ionization process at higher laser energies (>8.4 GW/cm²/cm⁻¹) and record laser energy dependencies of ~2 for H⁺ and D⁺. Although the dissociation energy of H₂ is ~4.5 eV and the energy of a photon at 243 nm is ~5.2 eV, absorption of a single photon is insufficient for the photoproduction of an H atom from H₂ because of negligible single-photon absorption at this wavelength (the ionization potentials of H₂ and O₂ are 12.06 and 15.43, respectively). Laser power dependence measurements for the photolysis of cold

flows of H₂ at atmospheric pressure were made using the optogalvanic probe. The laser beam was focused ~0.5 mm from the 1-mm-diameter anode of the probe, which was inserted in the flowing gas, H₂. Photolysis of H₂ produces ground state H atoms which are photoexcited and photoionized. The liberated electrons are then detected with the optogalvanic probe. A laser power dependence measurement gave a value of n=4.54+/-0.3. This result also suggests a two-photon dissociation of H₂ to produce H atoms. This reasoning is consistent with a two-photon dissociation of H₂ to produce ground state atoms, two-photon resonance excitation and a one photon ionization. These results, when compared to the time-of-flight data and variable pressure cell data, strongly suggest a collisional induced photodissociation process may occur.

This spectroscopic data gives valuable information on the laser power dependence for microplasma formation. The overall laser energy dependence for the microplasma formation must then be at least a five photon process and clearly is highly nonlinear. The absorption of two laser photons are required for the photoproduction of ground state atoms and the absorption of three additional laser photons are necessary for photoionization. In comparison, photoproduction of atomic oxygen from molecular oxygen photodissociation near 225.5 nm gave a two-photon dependence even though the single photon energy would be sufficient for photolysis (overall a five photon process) (Forch and Miziolek 1987).

3.3 Ignition Experiments. We then began a series of experiments on premixed flows of H_2/O_2 and D_2/O_2 at atmospheric pressure in order to investigate the possibility of using these microplasmas as ignition sources. Previous work (orch and Miziolek 1986, 1987; Syage et al. 1988; Lewis von Elbe 1951) has shown that the minimum ignition energy for these reactive gaseous mixtures occurs in the fuel lean region ($\Phi = 0.7$ –.8) and is characteristic of light, diffusive fuels such as H_2 , which can replenish burned fuel early on in the evolution of the ignition kernel as it expands into a stabilized flame front. In our recent work on excimer laser (ArF, 193 nm) ignition of H_2/O_2 we also found that the most efficient ignition was in the fuel lean region (Forch et al. 1989). In two separate, but identical experiments, premixed flows of either H_2/O_2 or D_2/O_2 which were held constant at $\Phi = 0.7$ were produced. The laser wavelength was varied and the amount of ILE required to ignite either flow into a stabilized combustion was measured. Figures 7a and 7b give the wavelength dependence on the amount of ILE necessary to ignite premixed flows of H2/O2 and D_2/O_2 , respectively. The

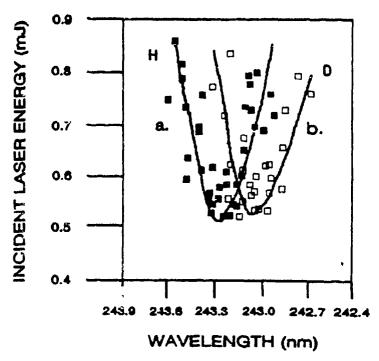


Figure 7. <u>ILE Necessary to Ignite Premixed Flows of: (a) H₂/O₂ and (b) D₂/O₂, as a Function of Excitation Wavelength Near 243 nm. A Shift of +11 cm⁻¹ of Ignition Curve b Relative to Ignition Curve a Is Evident.</u>

ignition curves clearly show a strong dependence of the ILE on the laser wavelength. Two prominent minima are evident. A simple comparison with the microplasma data (as described above) show that they correspond exactly to the spectral locations of the H and D atom twophoton resonance excitation wavelengths near 243 nm. Apparently, the focused UV laser near 243 nm not only photodissociates H₂ or D₂ to yield ground state atoms, but also, when on resonance with H or D, requires the least amount of laser energy to ignite the gases into a stabilized combustion. We believe that the ignition of these reactive gases occurs through the resonant formation of a microplasma from multiphoton excitation of H and D atoms. The wavelength shift of +11 cm-1 of the D_2/O_2 ignition curve relative to that of the H_2/O_2 curve is clearly related to the H-D deuterium isotope shift and underscores the importance of resonance enhancement in the ignition event. We next measured the dependence of the ILE (at a fixed laser wavelength) on Φ. Figure 8a and b gives plots of incident laser energy required for ignition of premixed flows of H₂/O₂ and D₂/O₂ vs. Φ when the laser is tuned to the two-photon resonance exitation wavelengths for H and D atoms, respectively. The observed minima in both curves occurs in the fuel lean region ($\Phi = 0.7$) and viith ILE=0.53+/-0.07 mJ and 0.52+/-0.05 for H_2/O_2 and D_2/O_2 , respectively. These experiments support the three step mechanism which we have proposed which consists of the laser photoproduction of ground

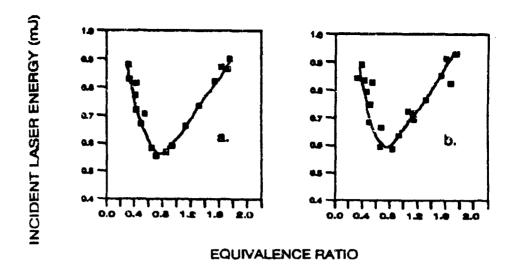


Figure 8. Dependence of the ILE Required to Ignite Premixed Flows of: (a) H₂/O₂ and (b) D₂/O₂, as a Function of Equivalence Ratio. The Laser Was Set at the Peak of the Two-Photon Excitation Wavelength of: (a) H Atoms at 243.07 and (b) D Atoms at 243.00.

state atoms, resonant multiphoton ionization of the atoms to liberate free electrons early on in the laser pulse and the formation of a laser produced microplasma which serves as an ignition kernel. We have also found that microplasmas could be formed very easily through resonance multiphoton ionization, at very low laser energy, that were not intense enough to cause ignition. Thus, our method allows for precise control of laser energy in the ignition process. By comparison, we have investigated ignition of these gas mixtures using the fundamental wavelength of the same Nd:YAG laser (1064 nm) or its second harmonic (532 nm) under identical experimental conditions and found that: 1) a factor of greater than forty times more laser energy was required for ignition; 2) ignition curve plots of ILE vs. Φ were flat (independent of Φ ; 3) a relatively intense spark was produced. This behavior is consistent with our mechanism and reasoning. The 1064 nm and 532 nm beams are essentially initially transparent to the reactive gaseous mixture (there is no absorption of the laser light). As the laser pulse grows in intensity with time, the laser field strength reaches a point where multiphoton ionization of the fuel and/or oxidizer liberates free electrons in the laser focal volume and a very intense laser spark is produced (through nonresonant gas breakdown). Here the laser energy greatly exceeds the requisite minimum ignition energy.

Visual examination of the laser ignition sparks produced from resonant multiphoton ionization and nonresonant gas breakdown are differ dramatically the former being much less intense than the latter. This is only a qualitiative observation at present, but we have purchased a high-speed streak camera to follow the temporal evolution of the laser-produced microplasmas into an ignition kernel and stabilized combustion and also perform more detailed spectroscopic examination of the sparks.) In our experiments, the incident laser energy, which is required for ignition, has been measured. Attempts to measure the energy absorbed by the microplasmas (through time-resolved absorption measurements) have been very difficult. The microplasma is heated to a very high temperature and, of course, expands with time wherein radiation losses could be significant. Furthermore, scattering of laser radiation by the microplasma occurs and has been calculated to be about 10% (Syage et al. 1988). The microplasmas are formed during the laser pulse (ca. / nsec) through a highly nonlinear process (at least a five photon dependence). Thus, it was very difficult to obtain reliable absorption data although recently we have successfully performed time-resolved absorption measurements on C2H2/O2 and C2H2/air ignition experiments. A rough stimate of the absorbed laser energy (making these corrections) is about 0.12 mJ at $\Phi = 0.7$ for H_2/O_2 and D_2/O_2 .

4. CONCLUSION

We have observed a strong wavelength dependence on the ignition of H_2/O_2 and D_2/O_2 premixed flows using a tunable laser near 243 nm. Furthermore, we have observed a wavelength shift in both microplasma formation in H_2 and D_2 gases and in the ignition of H_2/O_2 and D_2/O_2 flows. These results underscore the importance of two-photon atomic resonances in the microplasma formation process. We have demonstrated that our recent observations of a new laser ignition phenomenon that involves resonant multiphoton photochemical formation of microplasmas appears to be more general and applies to fuel molecules (H_2 and D_2) as well all exidizers (O_2 and N_2O). We believe that this is the first report of a sensitive wavelength dependence on the laser energy required to ignite these mixtures through resonant multiphoton excitation of H and D atoms (produced from H_2 and D_2 photolysis) and the first report of a deuterium isotope effect in laser ignition. Measurement of the ILE required for ignition vs. equivalence ratio (Φ) shows that the most efficient ignition occurred with 0.55 mJ iLE at $\Phi = 0.7$ in the fuel lean region. Strong experimental evidence is given which shows that ignition occurs through the resonant formation of a laser-produced

microplasma in a well-defined volume. These new experimental results indicate that resonance enhancement in the formation of a microplasma is a well controlled ignition method which appears to alleviate the problems associated with the sharp thresholds encountered in the well-known laser-produced spark (gas breakdown) process. Currently, we are exploring other possible resonance effects for the purpose of activating or enhancing the combustion of other reactive systems.

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